

**OLD BRIDGE CHEMICALS, INC.
OLD WATERWORKS ROAD
OLD BRIDGE, NJ 08857**

Telephone: 732-727-2225
Facsimile: 732-727-2653
info@oldbridgechem.com
www.oldbridgechem.com

MATERIAL SAFETY DATA SHEET

November 1, 2008

Page 1 of 6

Product Name: COPPER SULFATE CRYSTALS

Manufacturer: Old Bridge Chemicals, Inc.
P.O. Box 194
Old Bridge, New Jersey 08857

Telephone: (732) 727-2225
Emergency Telephone: (800) 275-3924
24-hour Emergency Telephone: (800) 424-9300 Chemtrec

HAZARD CLASSIFICATION: NFPA: Health-3 / Fire-0 / Reactivity-0
HMIS: Health-3 / Fire-0 / Reactivity-0

This document is prepared pursuant to the OSHA Hazard Communication Standard (29 CFR 1910.1200).

SECTION I. MATERIAL IDENTIFICATION

Common Name: Copper Sulfate
Synonyms: Blue Vitrol, Bluestone, Cupric Sulfate
Molecular Formula: $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
EPA Reg. Number: 46923-4
EPA Signal Word: **DANGER**
CAS Number: 7758-99-8
SIC Number: 28199 C 29

SECTION II. COMPOSITION/INFORMATION ON INGREDIENTS

Exposure Limits: ACGIH TLV TWA: 1.0 mg/m³ (as copper dust/mist)
OSHA PEL TWA: 1.0 mg/m³ (as copper dust/mist)

SECTION III. HAZARD INFORMATION

Emergency overview: Odorless blue crystals. Can cause irreversible eye damage and severe skin irritation. Harmful if swallowed or absorbed through the skin. Avoid

breathing dust or mist. Avoid contact with the skin, eyes or clothes. May cause skin sensitization in certain individuals.

Swallowing: Toxic orally in accordance with FHSLA regulations. Acute oral LD50 (male rats) = 472 mg/Kg. Can cause irritation to the digestive tract and abdominal pain.

Skin : Slight skin irritant. Excessive exposure may cause skin irritation. Repeated exposure may cause allergic dermatitis. May cause irritation or burns on wet skin.

Eyes: Can cause severe eye irritation and may result in irreversible eye damage.

Inhalation: Inhalation of dust may cause irritation to the mucous membranes and upper respiratory tract

Carcinogenicity: None as per NTP, OSHA, and IARC.

SECTION IV. FIRST AID PROCEDURES

Ingestion: Give large amounts of milk, egg white, gelatin solution, or if they are not available, large quantities of water. Do not induce vomiting or give anything to an unconscious person. Avoid alcohol. Call Poison Control Center or a physician.

Skin: Wash thoroughly with soap and water. Remove and wash contaminated clothing before reuse.

Eyes: Immediately flush eyes with plenty of water for 15 minutes. Hold eyelids apart during irrigation. Call a physician.

Inhalation: Remove person to fresh air. If not breathing, administer artificial respiration. Get medical attention.

Carcinogenicity: None.

SECTION V. FIRE AND EXPLOSION DATA

Flash Point: Not applicable.

Flammable Limits: Not flammable. If heated above 400° C it can decompose to emit toxic fumes of oxide and sulfur.

Auto Ignition Temperature: Not determined.

Extinguishing Media: Copper Sulfate does not burn nor will it support combustion. If stored with other combustible products use water, CO₂ or dry chemical.

Special Fire Fighting Instructions: If dry heated above 600° C, SO₂ is evolved. If water is used it will solubilize the Copper Sulfate and care should be taken to keep such water out of streams or other water bodies.

Fire Fighting Equipment: Wear self-contained breathing apparatus.

Fire and Explosion Hazards: None.

SECTION VI. ACCIDENTAL RELEASE MEASURES

Use clean-up measures that avoid dust generation. Wear NIOSH or MSHA approved respirator if dust will be generated. Cover spill with absorbent material such as seeping compound or lime. Sweep up and put into an appropriate container for proper disposal in an approved method. Prevent accidental entry of solution into streams or other bodies of water. Shovel spills into plastic bags and seal with tape.

SECTION VII. HANDLING AND STORAGE

Signal Word:

DANGER

Handling Information:

Avoid breathing dust or mist. Sweep up crystals. Eye wash stations should be available in work areas. Users should wash hands before eating, drinking, smoking or using the toilet. Remove PPE immediately after handling this product. Wash outside of gloves before removing. Wash and change into clean clothing as soon as possible.

Storage Information:

Store in closed containers in a cool, dry, well-ventilated area away from heat sources and reducing agents. Store in original containers. Keep away from galvanized pipe, aluminum and nylon. Place damaged containers in plastic bags.

SECTION VIII. EXPOSURE CONTROLS/PERSONAL PROTECTION

Ventilation:

TWA = 1 mg/l. for Copper Sulfate. When TWA exceeds this limit in the workplace, provide appropriate ventilation.

Respiratory Protection:

Wear an approved respirator for dusts or mists: MSHA/NIOSH approved number prefix TC-21C, or a NIOSH approved respirator with any R, P or HE filter. Alternatively, provide respiratory protection equipment in accordance with Paragraph 1910.134 of Title 29 of the Code of Federal Regulations.

Eye Protection:

Use safety glasses with side shields or goggles.

Skin Protection:

Applicators and other handlers should wear long-sleeved shirts and long pants, waterproof gloves, shoes plus socks and protective eyewear. Discard clothing and other absorbent materials that have been drenched or heavily contaminated with product's concentrate. Do not reuse them. Keep and wash PPE separately from other laundry.

SECTION IX. PHYSICAL DATA

Physical State:

Blue crystals or powder

Boiling Point:

-5H₂O @ 150°C

Melting Point:	-4H ₂ O @ 110°C
Vapor Density:	NA
Specific Gravity:	2.284
Solubility in H ₂ O:	22.37% @ 0°C 117.95% @ 100°C
Solubility in Other Solvents:	Soluble in methanol, glycerol and slightly soluble in ethanol.
Molecular Weight:	249.68
Appearance:	Transparent blue crystals
Odor:	Odorless

SECTION X. REACTIVITY DATA

Stability:	Stable
Conditions to Avoid:	Product is highly soluble, but does not react with water.
Incompatibility:	Solutions are mildly corrosive to steel. Store solutions in plastic or rubber or 304, 347 or 316 stainless steel. Iron and moisture should be avoided. Store in a dry area. Incompatible with aluminum powder, acetylene gas, hydroxylamine, magnesium and moisture. Contact with magnesium can generate dangerous levels of hydrogen gas. With exposure to air it will oxidize and turn whitish.
Hazardous Decomposition Products:	None at normal production temperatures and pressures. If dry heated above 600°C toxic sulfur may evolve.
Polymerization:	Will not occur.

SECTION XI. TOXICOLOGICAL INFORMATION

Inhalation:	LC ₅₀ > 2.95 mg/L (rat) Inhalation of dust can result in irritation of nasal mucous membranes and sometimes of the pharynx, on occasion ulceration with perforation of the nasal septum.
Skin:	LD ₅₀ > 8.0 g/kg (rabbit)
Ingestion:	LD ₅₀ > 472.5 mg/kg (rat)
Primary Eye Irritation:	Corrosive, irreversible eye damage.
Primary Skin Irritation:	No skin irritation.

Note to Physician: Probable mucosal damage may contraindicate the use of gastric lavage. Measures against circulatory shock, respiratory depression and convulsions may be needed. Wilson's disease can be aggravated by excessive exposure. Symptoms include nausea, vomiting, epigastria pain, diarrhea, dizziness, jaundice, and general debility.

SECTION XII. ENVIRONMENTAL AND DISPOSAL INFORMATION

Aquatic LC ₅₀ :	Daphnia magna 0.182 mg/L.
----------------------------	---------------------------

	Rainbow Trout 0.17 mg/L. Blue Gill 1.5 mg/L.
Spills and Leaks:	All values are expressed as Copper Sulfate Pentahydrate. Test water was soft. Period 24 hours. Comply with Federal, State and local regulations on reporting spills. Do not wash away crystals or powder. Recover dry if possible. If product is in a confined solution, react with soda ash to form an insoluble Copper Carbonate solid that can be scooped up.
Waste Disposal:	Do not reuse container. Comply with Federal, State and local environmental control regulations. Sweep up crystals, powder or insoluble Copper Carbonate and dispose of in an approved landfill.
Environmental Effects:	May be dangerous if it enters the public water systems. Follow local regulation. Toxic to fish and plants. Fish toxicity critical concentration is 235 mg/l. and plant toxicity is 25 mg/l.

SECTION XIII. REGULATORY INFORMATION

NOTICE: The information herein is presented in good faith and believed to be accurate. However, no warranty, express or implied, is given. Regulatory requirements are subject to change and may differ from one location to another. It is the buyer's responsibility to ensure that its activities comply with Federal, State, and local laws.

SARA 313 Information; This product contains the following substance subject to the reporting requirements of Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR Part 372: COPPER COMPOUND > 1.0%

SARA HAZARD CATEGORY: This product has been reviewed according to the EPA "Hazard Categories" promulgated under Sections 311 and 312 of the Superfund Amendment and Reauthorization Act of 1986 (SARA Title III) and is considered, under applicable definitions, to meet the following category:

ACUTE HEALTH HAZARD

OSHA: This product is considered hazardous under the OSHA Hazardous Communication Standard (29 CFR 1910.1200).

TSCA: Listed on the Chemical Inventory

CERCLA Hazardous Substances: RQ is not assigned to the broad class of copper compounds.

RCRA: When discarding this material as supplied, it does not meet RCRA characteristic definition if ignitability, corrosiveness, reactivity, and is not listed in 40CFR 261.33.

This product contains Copper Sulfate and is subject to the reporting requirements of Section 13 of the Emergency Planning and Community-Right-to-Know-Act of 1986 (40CFR372).

NSF: Maximum use in potable water applications per NSF formulation - 2 ppm

SECTION XIV. SHIPPING INFORMATION

DOT Shipping Name RQ, Environmentally Hazardous Substance, Solid, NOS, (CUPRIC SULFATE), 9, UN3077, PGIII, ERG 171

Reportable Quantity (RQ): 10 pounds (4.54 kg)

Not hazardous when shipping less than 10 pounds

Note: Marine pollutant

SECTION XV. MISCELLANEOUS INFORMATION

This is an NSF Certified Product to ANSI/NSF 60. Maximum use in potable water is not to exceed 2 mg. per liter.

SECTION XVI. MSDS PREPARATION INFORMATION

Prepared By: Joel L. Goldschmidt
Vice President